

For the Formula Shown, Write the Correct Name

ZnS - zinc sulfide (<i>No Roman Numeral</i>)	CuCl ₂ - copper(II) chloride
MgCl ₂ - magnesium chloride	PCl ₅ - phosphorous pentachloride
Ca(ClO ₃) ₂ - calcium chlorate	LiNO ₂ - lithium nitrite
CaSO ₄ - calcium sulfate	KH ₂ PO ₄ - potassium dihydrogen phosphate
AgNO ₃ - silver nitrate	CuCN - copper(I) cyanide
H ₂ S(aq) - hydrosulfuric acid	KHCO ₃ - potassium hydrogen carbonate
CaO - calcium oxide	NaHSO ₄ - sodium hydrogen sulfate
H ₂ CO ₃ (aq) - carbonic acid	Li ₂ HPO ₄ - lithium hydrogen phosphate
Mg ₃ (PO ₄) ₂ - magnesium phosphate	H ₃ PO ₃ (aq) - phosphorous acid
KCl - potassium chloride	MgSO ₄ - magnesium sulfate
K ₂ O - potassium oxide	Ca(IO ₂) ₂ - calcium iodite
Al(NO ₂) ₃ - aluminum nitrate	SiO ₂ - silicon dioxide
MgO - magnesium oxide	CuCl - copper(I) chloride
SnI - tin(I) iodide	KClO ₂ - potassium chlorite
AsCl ₅ - arsenic pentachloride	CaSO ₃ - calcium sulfite
CuSO ₃ - copper(II) sulfite	NaBr - sodium bromide
HF(aq) - hydrofluoric acid	P ₂ O ₃ - diphosphorous trioxide
FeSO ₄ - iron(II) sulfate	HClO(aq) - hypochlorous acid
SnCl ₄ - tin(IV) chloride	NO ₂ - nitrogen dioxide
AsCl ₃ - arsenic trichloride	NaH - sodium hydride
KCN - potassium cyanide	ZnS - zinc sulfide (<i>No Roman Numeral</i>)
NH ₄ OH - ammonium hydroxide	Pb(NO ₃) ₂ - lead(II) nitrate
Fe(ClO ₄) ₃ - iron(III) perchlorate	H ₂ Se(aq) - hydroselenic acid
HNO ₂ (aq) - nitrous acid	H ₃ PO ₄ (aq) - phosphoric acid
CS ₂ - carbon disulfide	CaH ₂ - calcium hydride